

5.00 credits

30.0 h + 30.0 h

Q2

Teacher(s)	Garcia Yann ;Leysens Tom ;
Language :	French
Place of the course	Louvain-la-Neuve
Prerequisites	<p>It is recommended to have acquired the knowledge and skills developed in the teaching units:</p> <p>LCHM1111 Chimie générale LPHY1101 Physique 1 LPHY1102 Physique 2</p> <p><i>The prerequisite(s) for this Teaching Unit (Unité d'enseignement – UE) for the programmes/courses that offer this Teaching Unit are specified at the end of this sheet.</i></p>
Main themes	<p>0. INTRODUCTION TO PHYSICAL CHEMISTRY Chemical equilibrium and partition coefficients. Applications.</p> <p>1. THERMODYNAMICS First principle of thermodynamics. Thermochemistry. Second principle of thermodynamics. Free Enthalpy.</p> <p>2. PHASE EQUILIBRIA Generalities. One-component systems: state diagram of a pure body. Thermodynamics and phase transition temperature. Phase rule. Two-component systems: binary phase diagrams.</p> <p>3. CHEMICAL EQUILIBRIA IN SOLUTION A/ Complements of acid-base equilibria and pH-metry. B/ Solubility and complexation. Complex reaction networks. Quantitative study of some cases.</p> <p>4. COMPLEMENTS OF ELECTROCHEMISTRY Electrolysis. Conductivity of solutions. Batteries. <i>Part 3A is not included in LCHM1211A.</i></p>

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Programmes containing this learning unit (UE)

Program title	Acronym	Credits	Prerequisite	Learning outcomes
Bachelor in Bioengineering	BIR1BA	5	LBIR1140 AND LBIR1170	